***Mixed Lewis Structure practice***Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Draw the Lewis dot structures for each compound. Remember… ionic is a transfer of electrons between \_\_\_\_\_\_\_\_\_\_\_\_\_\_ and \_\_\_\_\_\_\_\_\_\_\_\_\_\_ while covalent is the sharing between \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.

1. H2O
2. CaF2
3. Al2O3
4. CH4
5. N2
6. NH3
7. CoCl2
8. Br2
9. BeO
10. ZnBr2