

Periodic Table of the Elements

1 IA 11A H Hydrogen 1.008																	2 IIA 2A He Helium 4.003
3 Li Lithium 6.941	4 Be Beryllium 9.012											5 B Boron 10.811	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180
11 Na Sodium 22.990	12 Mg Magnesium 24.305	3 IIIB 3B	4 IVB 4B	5 VB 5B	6 VIB 6B	7 VIIB 7B	8 VIII 8	9 VIII 8	10 VIII 8	11 IB 1B	12 IIB 2B	13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.066	17 Cl Chlorine 35.453	18 Ar Argon 39.948
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.933	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.39	31 Ga Gallium 69.732	32 Ge Germanium 72.61	33 As Arsenic 74.922	34 Se Selenium 78.09	35 Br Bromine 79.904	36 Kr Krypton 84.80
37 Rb Rubidium 84.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.907	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.906	46 Pd Palladium 106.42	47 Ag Silver 107.868	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.71	51 Sb Antimony 121.760	52 Te Tellurium 127.6	53 I Iodine 126.904	54 Xe Xenon 131.29
55 Cs Cesium 132.905	56 Ba Barium 137.327	57-71	72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.85	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.08	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.383	82 Pb Lead 207.2	83 Bi Bismuth 208.980	84 Po Polonium [208.982]	85 At Astatine 209.987	86 Rn Radon 222.018
87 Fr Francium 223.020	88 Ra Radium 226.025	89-103	104 Rf Rutherfordium [261]	105 Db Dubnium [262]	106 Sg Seaborgium [266]	107 Bh Bohrium [264]	108 Hs Hassium [269]	109 Mt Meitnerium [268]	110 Ds Darmstadtium [269]	111 Rg Roentgenium [272]	112 Cn Copernicium [277]	113 Uut Ununtrium unknown	114 Fl Flerovium [289]	115 Uup Ununpentium unknown	116 Lv Livermorium [298]	117 Uus Ununseptium unknown	118 Uuo Ununoctium unknown

Lanthanide Series	57 La Lanthanum 138.906	58 Ce Cerium 140.115	59 Pr Praseodymium 140.908	60 Nd Neodymium 144.24	61 Pm Promethium 144.913	62 Sm Samarium 150.36	63 Eu Europium 151.966	64 Gd Gadolinium 157.25	65 Tb Terbium 158.925	66 Dy Dysprosium 162.50	67 Ho Holmium 164.930	68 Er Erbium 167.26	69 Tm Thulium 168.934	70 Yb Ytterbium 173.04	71 Lu Lutetium 174.967
Actinide Series	89 Ac Actinium 227.028	90 Th Thorium 232.038	91 Pa Protactinium 231.036	92 U Uranium 238.029	93 Np Neptunium 237.048	94 Pu Plutonium 244.064	95 Am Americium 243.061	96 Cm Curium 247.070	97 Bk Berkelium 247.070	98 Cf Californium 251.080	99 Es Einsteinium [254]	100 Fm Fermium 257.095	101 Md Mendelevium 258.1	102 No Nobelium 259.101	103 Lr Lawrencium [262]

Table 14-1 Oxidation Numbers/ Charges of Some Elements and Polyatomic Ions

Charge +1	
H ⁺	hydrogen
Au ⁺	gold (I) or aurous
Ag ⁺	silver
Cu ⁺	copper (I) or cuprous
Hg ₂ ²⁺	mercury (I) or mercurous
K ⁺	potassium
Na ⁺	sodium
Li ⁺	Lithium
NH ₄ ⁺	ammonium
H ₃ O ⁺	hydronium
Tl ⁺	thallium (I)

Charge +2	
Ba ²⁺	barium
Be ²⁺	Beryllium
Pt ²⁺	Platinum (II) orPlatinous
Ca ²⁺	calcium
Cd ²⁺	Cadmium
Mg ²⁺	magnesium
Sr ²⁺	strontium
Zn ²⁺	zinc
Co ₂₊	cobalt (II) or cobaltous
Cu ²⁺	copper (II) or cuprous
Fe ²⁺	Iron (II) or ferrous
Pb ²⁺	lead (II) or plumbous
Mn ²⁺	manganese (II) or manganous
Hg ²⁺	mercury (II) or mercuric
Ni ²⁺	nickel (II) or nickelous
Sn ²⁺	tin (II) or stannous
Cr ²⁺	chromium (II) or chromous

Charge +3	
Al ³⁺	aluminum
Bi ³⁺	bismuth
B ³⁺	boron
As ³⁺	arsenic (III) or arsenious
Sb ³⁺	antimony (III) or antimonous
Co ³⁺	cobalt (III) cobaltic
Au ³⁺	gold (III) or auric
Fe ³⁺	iron (III) or ferric
Ga ³⁺	gallium
Tl ³⁺	thallium (III)
Cr ³⁺	chromium (III) or chromic

Charge +4	
Pb ⁴⁺	lead(IV) or plumbic
Pt ⁴⁺	Platinum (IV) or platinic
Mn ⁴⁺	Manganese (IV) or manganic
Sn ⁴⁺	Tin (IV) or stannic

Charge -1	
F ⁻	fluoride
Cl ⁻	chloride
Br ⁻	bromide
I ⁻	iodide
ClO ⁻	hypochlorite
ClO ₂ ⁻	chlorite
ClO ₃ ⁻	chlorate
ClO ₄ ⁻	perchlorate
NO ₃ ⁻	nitrate
NO ₂ ⁻	nitrite
CN ⁻	cyanide
C ₂ H ₃ O ₂ ⁻	acetate
HCO ₃ ⁻	bicarbonate or hydrogen carbonate
H ₂ PO ₄ ⁻	dihydrogen phosphate
MnO ₄ ⁻	permanganate
OH ⁻	hydroxide
SCN ⁻	thiocyanate
H ⁻	hydride
HSO ₄ ²⁻	bisulfate or hydrogen sulfate
HSO ₃ ²⁻	bisulfate or hydrogen sulfite
IO ₃ ⁻	iodate
BrO ₃ ⁻	bromate
BrO ⁻	hypobromite
HCO ₂ ⁻	formate

Charge -2	
O ²⁻	oxide
S ²⁻	sulfide
Se ²⁻	selenide
CO ₃ ²⁻	carbonate
Cr ₂ O ₇ ²⁻	dichromate
CrO ₄ ²⁻	chromate
SO ₄ ²⁻	sulfate
SO ₃ ²⁻	sulfite
SiO ₃ ²⁻	silicate
HPO ₄ ²⁻	hydrogen phosphate
O ₂ ²⁻	peroxide
C ₂ O ₄ ²⁻	oxalate
S ₂ O ₃ ²⁻	thiosulfate

Charge -3	
N ³⁻	nitride
P ³⁻	phosphide
AsO ₄ ³⁻	arsenate
PO ₃ ³⁻	phosphite
PO ₄ ³⁻	phosphate
As ³⁻	arsenide
BO ₃ ³⁻	borate

Metal Activity Series for Single Replacement Reactions

Metal	Metal Ion	Reactivity
Lithium	Li^+	Most Reactive
Potassium	K^+	↓
Calcium	Ca^{2+}	↓
Sodium	Na^+	↓
Magnesium	Mg^{2+}	↓
Aluminum	Al^{3+}	↓
Manganese	Mn^{2+}	↓
Zinc	Zn^{2+}	↓
Chromium	$\text{Cr}^{2+}, \text{Cr}^{3+}$	↓
Iron	$\text{Fe}^{2+}, \text{Fe}^{3+}$	↓
Tin	$\text{Sn}^{+2}, \text{Sn}^{+4}$	↓
Lead	Pb^{2+}	↓
Hydrogen	H^+	↓
Copper	Cu^{2+}	↓
Silver	Ag^+	↓
Mercury	Hg^{2+}	↓
Platinum	Pt^{2+}	↓
Gold	$\text{Au}^+, \text{Au}^{3+}$	Least Reactive

Solubility Chart for Double Replacement Reactions

	Bromide Br ⁻	Carbonate CO ₃ ⁻²	Chloride Cl ⁻	Hydroxide OH ⁻	Nitrate NO ₃ ⁻	Oxide O ⁻²	Phosphate PO ₄ ⁻³	Sulfate SO ₄ ⁻²
Aluminum Al ⁺³	S	X	S	I	S	I	I	S
Ammonium NH ₄ ⁺	S	S	S	S	S	X	S	S
Calcium Ca ⁺²	S	I	S	S	S	SS	I	SS
Copper (II) Cu ⁺²	S	X	S	I	S	I	I	S
Iron (II) Fe ⁺²	S	I	S	I	S	I	I	S
Iron (III) Fe ⁺³	S	X	S	I	S	I	I	SS
Magnesium Mg ⁺²	S	I	S	I	S	I	I	S
Potassium K ⁺	S	S	S	S	S	S	S	S
Silver Ag ⁺	I	I	I	X	S	I	I	SS
Sodium Na ⁺	S	S	S	S	S	S	S	S
Zinc Zn ⁺²	S	I	S	I	S	I	I	S

Key:

S = soluble	I = insoluble
SS = slightly soluble	X = other